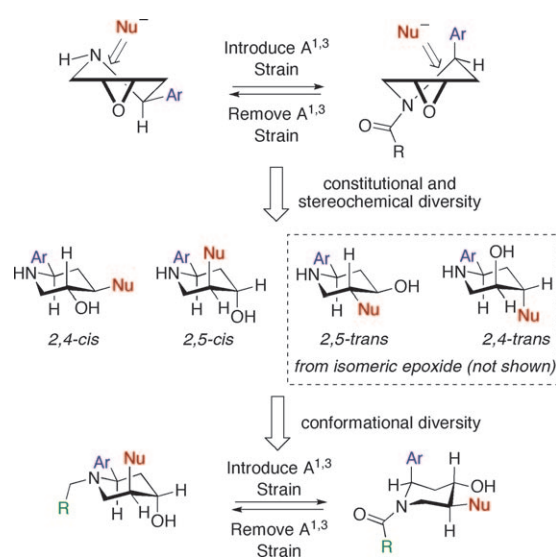


1,3-Allylic Strain as a Strategic Diversification Element for Constructing Libraries of Substituted 2-Arylpiperidines**

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Screening approaches to probe molecules for drug discovery require access to high-quality, small-molecule libraries. One contemporary challenge in providing such access is the construction of libraries that maximize the coverage of chemical (functional group), stereochemical, and spatial diversity in a given chemotype.^[1] Although the problem of functional group diversity has been addressed since the earliest days of combinatorial chemistry and parallel synthesis, the incorporation of stereochemical diversity and, more broadly, shape diversity has required the development of new strategies. These include the use of spatially diverse scaffolds or the pre-construction of stereochemically diverse building blocks that are then combined to afford final products ("build-couple-pair"^[2] is an example of this).

Herein we describe a conformational switching approach toward shape-diverse piperidine libraries in which the presence or absence of 1,3-allylic strain ($A^{1,3}$ strain)^[3] is leveraged to enhance both 1) scaffold diversity by regiodivergent opening of epoxide intermediates and 2) the conformational space of the final library through the simple expedient of changing the nature of nitrogen substitution.^[4] The concept is illustrated in Scheme 1 for a series of substituted 2-arylpiperidines. For a given epoxide isomer, the conformation of the piperidine ring will depend on whether the N1 atom is sp^3 hybridized (Ar preferring an equatorial position because of the minimization of 1,3-diaxial interactions) or sp^2 (Ar preferring axial position because of $A^{1,3}$ strain in the equatorial isomer).^[5] Nucleophilic addition to the epoxide would then take place according to the Fürst-Plattner principle^[6] (*trans*-diaxial opening), leading to two constitutional isomers from this epoxide intermediate (2,4- versus 2,5-*cis* Ar/Nu relationships). *Stereochemical diversity* would then follow by applying the same principles to the alternative



Scheme 1. Use of $A^{1,3}$ strain to control constitutional (2,4- vs. 2,5-Ar/Nu relationship), stereochemical (*cis* vs. *trans*), and conformational diversity in piperidine libraries. Functional group diversity arises from variations in Ar, Nu, and R.

epoxide diastereomer, affording the analogous 2,4- and 2,5-*trans* isomers. Once prepared—and likely following the downstream introduction of *functional group diversity*—the compounds in the library could then be substituted at N by a different set of alkyl or acyl substituents, thus leading to a doubling of the library members through *conformational diversity*.

We chose to demonstrate this approach by preparing a library based on the triazole-containing piperidines shown in Scheme 2 (selected because the arylpiperidine chemotype appears in a number of bioactive compounds and is therefore a desirable library scaffold for broad screening).^[7] To a first approximation, the expected conformations in one such library are shown (four isomers bearing two different N groups), thus demonstrating the range of conformational and configurational space covered by these compounds.

To demonstrate the value of 1,3-allylic strain in scaffold preparation, we first carried out the stereochemically and constitutionally differentiated scaffold syntheses shown in Scheme 3. Four 2-aryl-1,2,3,6-tetrahydropyridines were constructed from substituted benzaldehydes^[8] in two steps by bisallylation/N-acylation and subsequent ring-closing metathesis. The N-acylated derivatives underwent highly stereoselective epoxidation reactions, presumably because the top faces as drawn were blocked by the aromatic groups in the most stable conformations.^[5] By using *m*-CPBA or methyl-

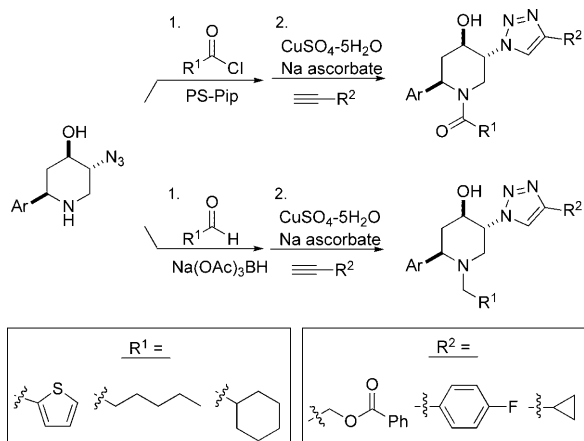
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Scheme 4. Scaffold diversification. PS-Pip = polystyrene-based piperidine.

isomer adopted the same conformation in solution, regardless of the substituents appended to the core. The same held true for the corresponding amides.

The conformational profiles of the eight stereochemical families of the triazole-containing piperidines (4 amines and 4 amides) are shown in Figure 1.^[14] Figure 1 a shows overlays of each amino/piperidine (blue) paired with the corresponding amido/piperidine (green) counterpart, highlighting the conformational differences achieved through the introduction of 1,3-allylic strain. Three of the four families of amino/piperidines adopted chair conformations, thus placing the 2-aryl substituents in the equatorial position. The 2,4-*trans* family of amino/piperidines exhibited a slight distortion from an ideal chair conformation, while still placing the 2-aryl substituents in a pseudoequatorial position. The corresponding 2,5-*cis* and 2,4-*trans* families of amido/piperidines adopted the opposite chair conformations, thus placing the 2-aryl substituents in the axial position. However, rather than adopting chair conformations and placing all three non-hydrogen substituents (Ph, OH, triazole) in axial positions, the 2,5-*trans* amido/piperidines exhibited twistlike conformations, and the 2,4-*cis* amido/piperidines adopted boat conformations. Thus, each of the eight compounds in the family presents the piperidine substituents in a unique three-dimensional array. Taken together, these compounds (and by extension the entire library that they represent) comprise a shape-diverse collection with predictable three-dimensional shapes for use in biological screening and the development of structure–activity relationship (SAR) studies (Figure 1 b).

Through these efforts, we have demonstrated a useful protocol for maximizing the stereochemical diversity in a piperidine library using a limited number of scaffolds and building blocks. This approach features the use of 1,3-allylic strain for controlling both the ring opening of epoxide precursors and, thus, constitutional isomerism, as well as the conformations of the final library members. In so doing, a library of 268 druglike compounds having predictable conformations has been prepared. The compounds prepared in this work are being scrutinized by high-throughput screening whereas the concepts utilized in the present case are currently

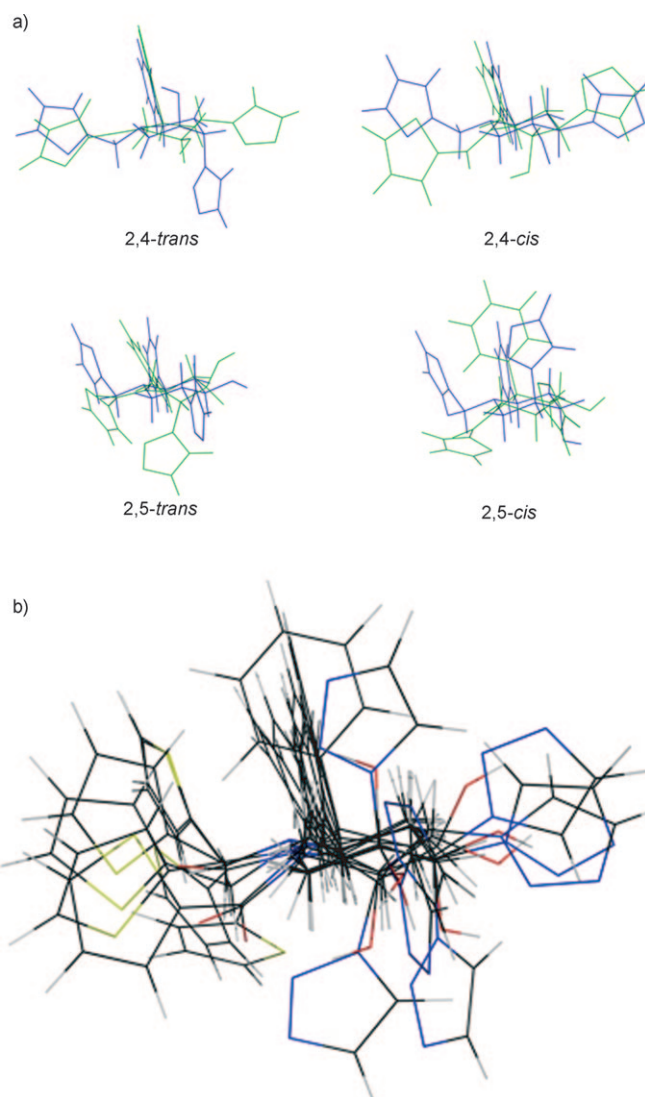


Figure 1. a) Overlays contrasting amino/piperidine conformation (blue) with the corresponding amido/piperidine conformation (green) for each isomeric pair (NMR). b) Overlay showing the chemical space occupied by the entire library. For each case: Ar = Ph, R¹ = 2-thiophene, R² = CH₂OBz. For individual conformational analyses, see the Supporting Information. Bz = benzoyl.

being applied to the construction of other heterocyclic libraries.

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